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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=294 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.046$
$w R$ factor $=0.127$
Data-to-parameter ratio $=12.6$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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# 2,6-Diamino-4-(4-methoxyphenyl)-1,4-dihydro-pyridine-3,5-dicarbonitrile $N, N$-dimethylformamide solvate 

The title compound, $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{~N}_{5} \mathrm{O} \cdot \mathrm{C}_{3} \mathrm{H}_{7} \mathrm{NO}$, was synthesized by the reaction of 4-methoxybenzaldehyde with malononitrile and ammonium acetate under microwave irradiation. The dihedral angle between the pyridine and benzene planes is 56.07 (8) ${ }^{\circ}$.

## Comment

3,5-Dicyanopyridine derivatives exhibit a wide range of bioactivities, such as antifungal, insecticidal, herbicidal, miticidal, nematocidal and antimicrobial activity (Gante \& Lust, 1971). Some substituted 3,5-dicyanopyridines have recently been reported to exhibit a high conductance-type calciumactivated K-channel opening effect (Hirochika et al., 2003) and to be adenosine receptor-selective ligands (Rosentreter et al., 2002), which are useful in the treatment of many diseases. More importantly, they are also versatile intermediates in organic synthesis (Castedo et al., 1984). As a consequence, much attention has been paid to the synthesis of these derivatives during the past 50 years. We report here the crystal structure of the title compound, (I).

(I)

The 2,6-diamino-1,4-dihydropyridine-3,5-dicarbonitrile fragment of (I) (atoms C8-C14/N1-N5) is almost planar, with an r.m.s. deviation of $0.076 \AA$ (Fig. 1). The dihedral angle between the pyridine and benzene planes is $56.07(8)^{\circ}$.

The crystal packing shows that intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{O}$, $\mathrm{N}-\mathrm{H} \cdots \mathrm{N}, \quad \mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds (Table 1) form a three-dimensional network (Fig. 2).

## Experimental

Compound (I) was prepared by the reaction of 4-methoxybenzaldehyde ( 1 mmol ) with malononitrile ( 2 mmol ) and ammonium acetate ( 1 mmol ) under microwave irradiation for 4 min (yield $95 \%$; m.p. 573 K). Single crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an ethanol and $N, N$-dimethylformamide solution (5:1 $\mathrm{v} / \mathrm{v}$ ).

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Figure 1
The structure (I), showing $30 \%$ probability displacement ellipsoids and the atom-numbering scheme.

## Crystal data

| $\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{~N}_{5} \mathrm{O} \cdot \mathrm{C}_{3} \mathrm{H}_{7} \mathrm{NO}$ | $Z=2$ |
| :---: | :---: |
| $M_{r}=338.37$ | $D_{x}=1.272 \mathrm{Mg} \mathrm{m}^{-3}$ |
| Triclinic, $P \overline{1}$ | Mo $K \alpha$ radiation |
| $a=6.9580$ (13) $\AA$ | Cell parameters from 1237 |
| $b=9.5297$ (18) $\AA$ | reflections |
| $c=14.695$ (3) $\AA$ | $\theta=2.3-25.6^{\circ}$ |
| $\alpha=99.900$ (3) ${ }^{\circ}$ | $\mu=0.09 \mathrm{~mm}^{-1}$ |
| $\beta=102.505$ (3) ${ }^{\circ}$ | $T=294$ (2) K |
| $\gamma=106.527(3)^{\circ}$ | Plate, colourless |
| $V=883.1$ (3) $\AA^{3}$ | $0.22 \times 0.18 \times 0.06 \mathrm{~mm}$ |
| Data collection |  |
| Bruker SMART CCD area-detector diffractometer | 3087 independent reflections 1774 reflections with $I>2 \sigma(I)$ |
| $\varphi$ and $\omega$ scans | $R_{\text {int }}=0.026$ |
| Absorption correction: multi-scan | $\theta_{\text {max }}=25.0^{\circ}$ |
| (SADABS; Sheldrick, 1996) | $h=-6 \rightarrow 8$ |
| $T_{\text {min }}=0.976, T_{\text {max }}=0.995$ | $k=-11 \rightarrow 11$ |
| 4506 measured reflections | $l=-14 \rightarrow 17$ |
| Refinement |  |
| Refinement on $F^{2}$ | $w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}{ }^{2}\right)+(0.0527 P)^{2}\right.$ |
| $R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.046$ | + 0.1375P] |
| $w R\left(F^{2}\right)=0.127$ | where $P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3$ |
| $S=1.00$ | $(\Delta / \sigma)_{\text {max }}=0.001$ |
| 3087 reflections | $\Delta \rho_{\text {max }}=0.18$ e $\AA^{-3}$ |
| 245 parameters | $\Delta \rho_{\text {min }}=-0.20 \mathrm{e}^{\AA^{-3}}$ |

H atoms treated by a mixture of independent and constrained refinement

Table 1
Hydrogen-bond geometry $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 3-\mathrm{H} 3 A \cdots \mathrm{O} 2^{\mathrm{i}}$ | $0.88(3)$ | $2.07(3)$ | $2.846(3)$ | $147(2)$ |
| $\mathrm{N} 3-\mathrm{H} 3 B \cdots \mathrm{~N} 4^{\text {ii }}$ | $1.00(3)$ | $2.04(3)$ | $3.033(3)$ | $168(2)$ |
| $\mathrm{N} 5-\mathrm{H} 5 A \cdots \mathrm{~N}^{\text {iii }}$ | $0.79(3)$ | $2.45(3)$ | $3.195(3)$ | $158(2)$ |
| $\mathrm{N} 5-\mathrm{H} 5 B \cdots \mathrm{O} 2^{\mathrm{iii}}$ | $0.98(3)$ | $2.13(3)$ | $3.083(3)$ | $166(2)$ |
| $\mathrm{C} 6-\mathrm{H} 6 A \cdots \mathrm{~N}^{\text {iv }}$ | 0.93 | 2.59 | $3.506(4)$ | 171 |
| $\mathrm{C} 7-\mathrm{H} 7 A \cdots \mathrm{O}^{\mathrm{v}}$ | 0.93 | 2.58 | $3.503(4)$ | 170 |
| $\mathrm{C} 17-\mathrm{H} 17 A \cdots \mathrm{~N} 1^{\text {vi }}$ | 0.93 | 2.48 | $3.288(4)$ | 145 |

Symmetry codes: (i) $x+1, y-1, z$; (ii) $-x+1,-y,-z+1$; (iii) $-x,-y+1,-z+1$; (iv) $x-1, y, z$; (v) $-x,-y+1,-z$; (vi) $x-1, y+1, z$.


Figure 2
The molecular packing of (I), viewed approximately along the $c$ axis. Dashed lines indicate hydrogen bonds.

The H atoms of the amino groups were located in a difference Fourier map and refined isotropically $[\mathrm{N}-\mathrm{H}=0.79$ (3)-1.00 (3) A$]$. All other H atoms were placed in idealized positions and allowed to ride on their parent atoms, with $\mathrm{C}-\mathrm{H}$ distances of 0.93 or $0.96 \AA$, and with $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}(\mathrm{C})$ for methyl H atoms and $1.2 U_{\text {eq }}(\mathrm{C})$ for other H atoms.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

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## References

Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
Castedo, L., Quintela, J. M. \& Riguera, R. (1984). Eur. J. Med. Chem. 19, 555557.

Gante, J. \& Lust, S. (1971). US Patent 3629270.
Hirochika, H., Ayako, M., Tomofumi, T., Toshio, O. \& Yusuke, H. (2003). Jpn. Patent 2003183254.
Rosentreter, U., Kraemer, T., Vaupel, A. \& Huebsch, J.-P. (2002). WO Patent 2002070520 A1.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

